

FLUORESCENCE

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. Uppsala University . Platform . Biological Visualization .



Light & Electron Microscopy : Flow Cytometry : Image Analysis

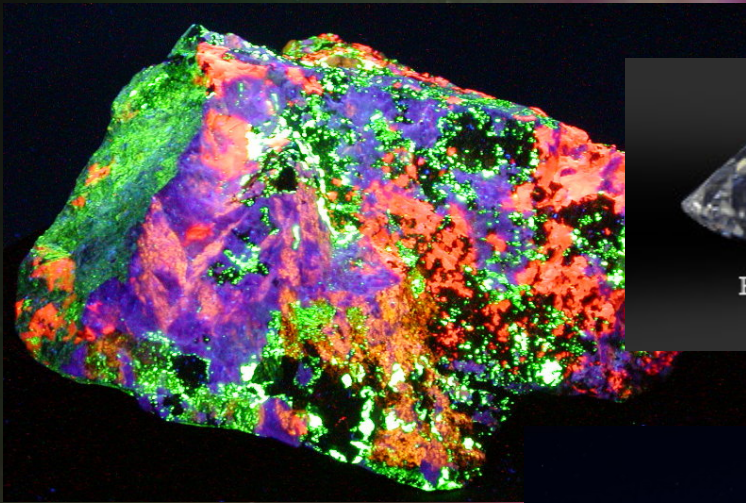
Information

This lecture contains images and information from the following internet homepages

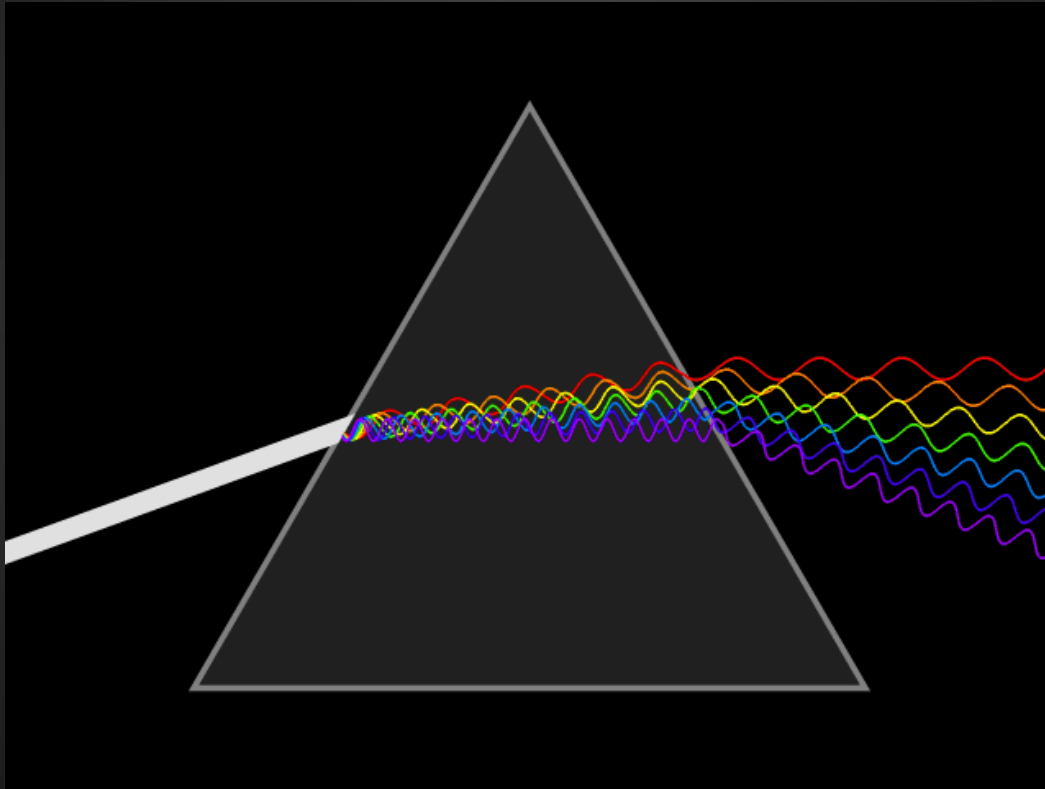
<http://micro.magnet.fsu.edu/primer/index.html>

<http://www.microscopyu.com/>

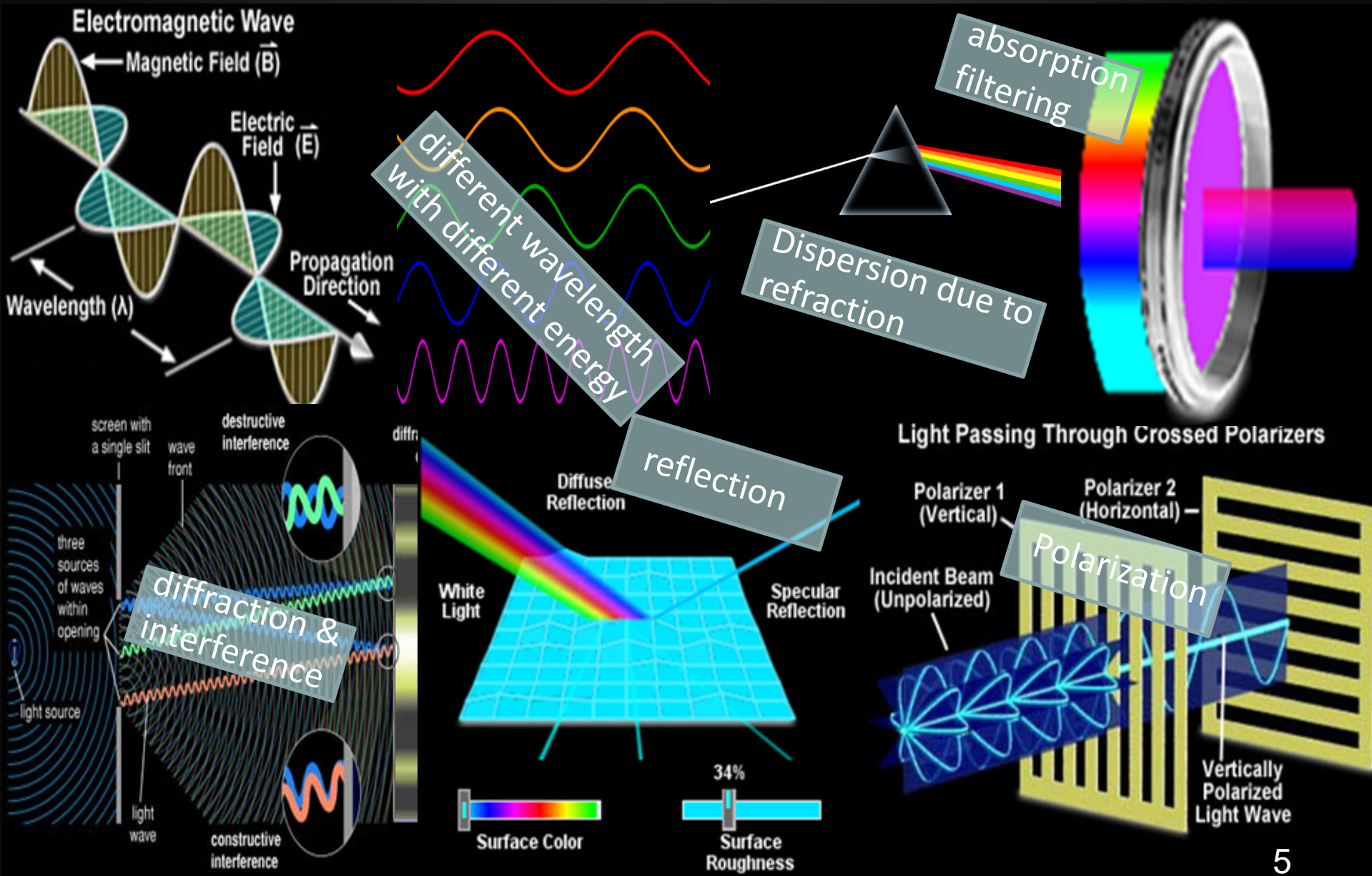
<http://www.olympusmicro.com/primer/lightandcolor/index.html>



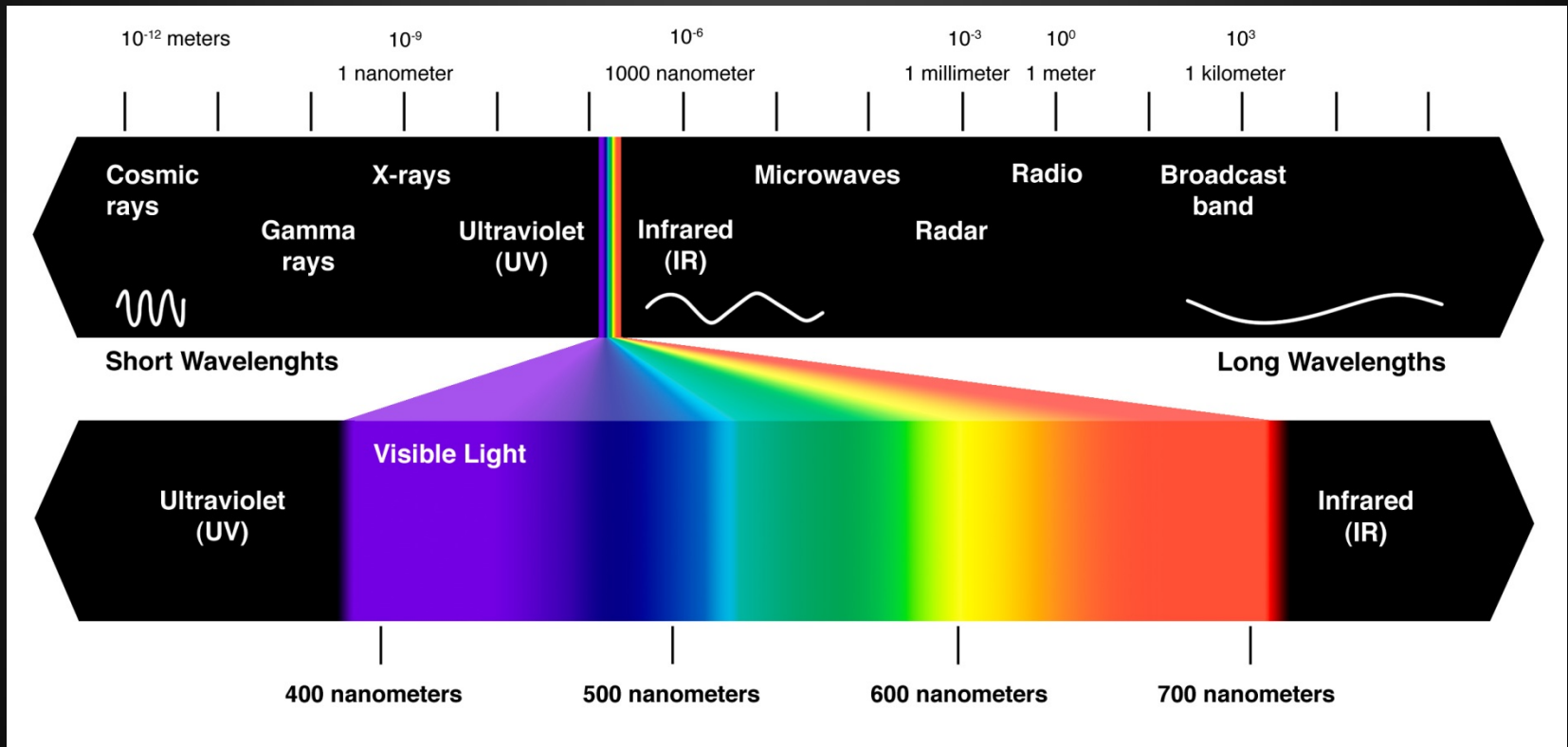
Light phenomenon



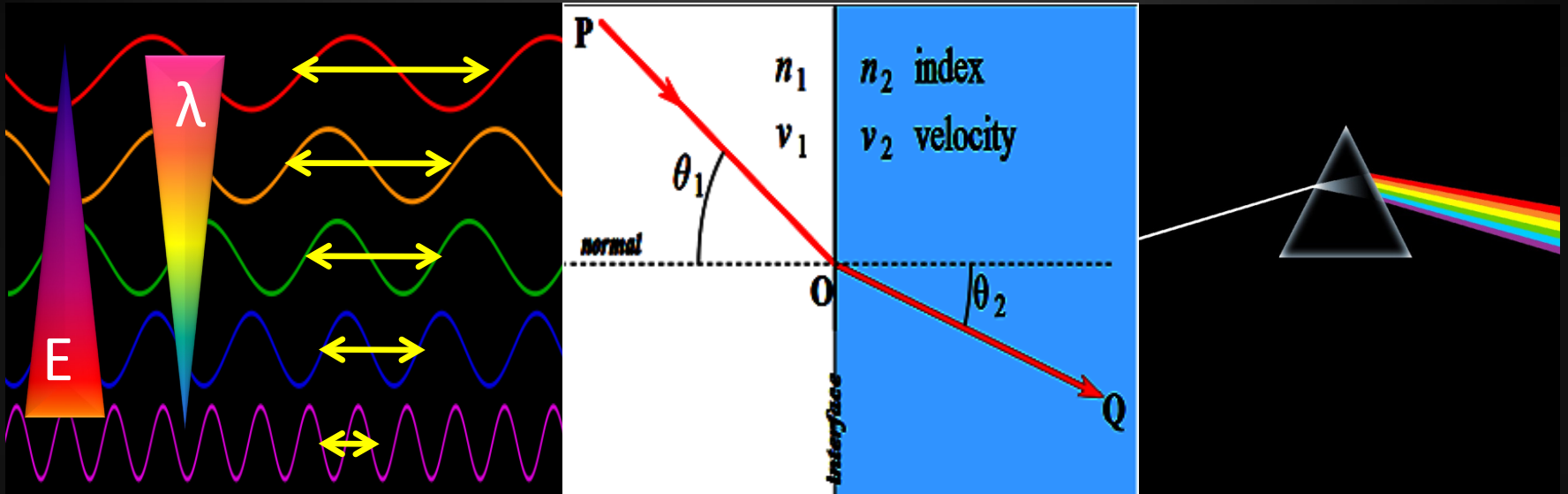
Wave or particle? ... Wave!



Wave or particle? ... Wave!



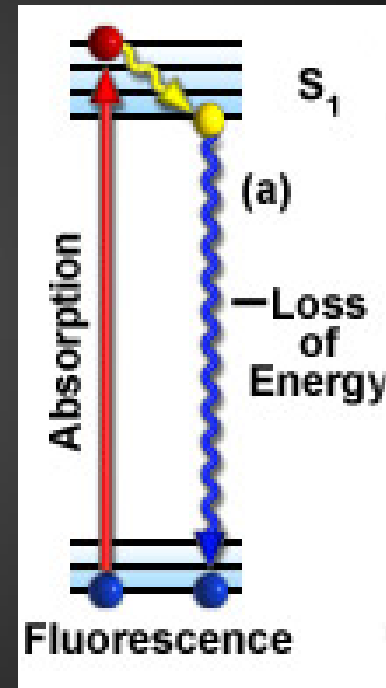
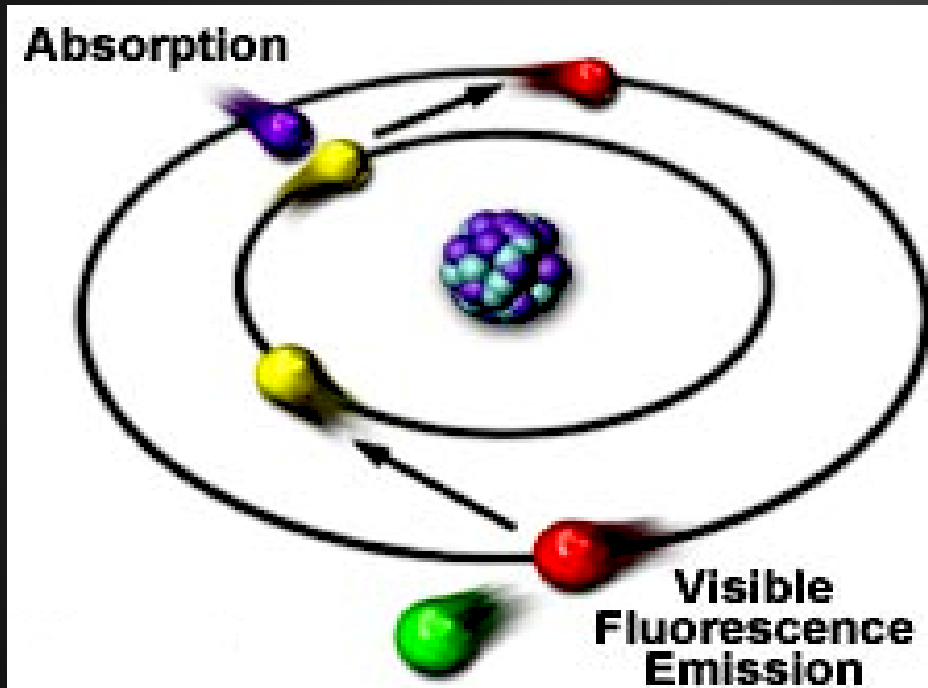
Wave or particle? ... Wave!



- Light travels in vacuum $C = 300.000 \text{ km/sec}$
- Seems to slows down in denser matter.
- → Refractive index $n = C/v$ ($v = \text{speed in matter}$)
- → light is 1.5 x slower at $n=1.5$
- → Light changes direction between different dense materials : shorter λ refract more than longer λ
- Optic lenses, effects in sample...
- shorter wavelength (blue) has higher energy

Wave or particle? ... Particle!

Principle of fluorescence



Light behaves sometimes as quantized energy pockets, light has particle behavior.

Fluorescence basics



Fluorescence basics

Definition:

Fluorescence is the emission of light by a substance that has absorbed light or other electromagnetic radiation.

Usually the emitted light has a longer wavelength, and therefore lower energy than the absorbed radiation.

Emission of light happens in time scale of nano second – so to speak immediately

Compared to **Phosphorescence:**

- specific type of photoluminescence related to fluorescence.

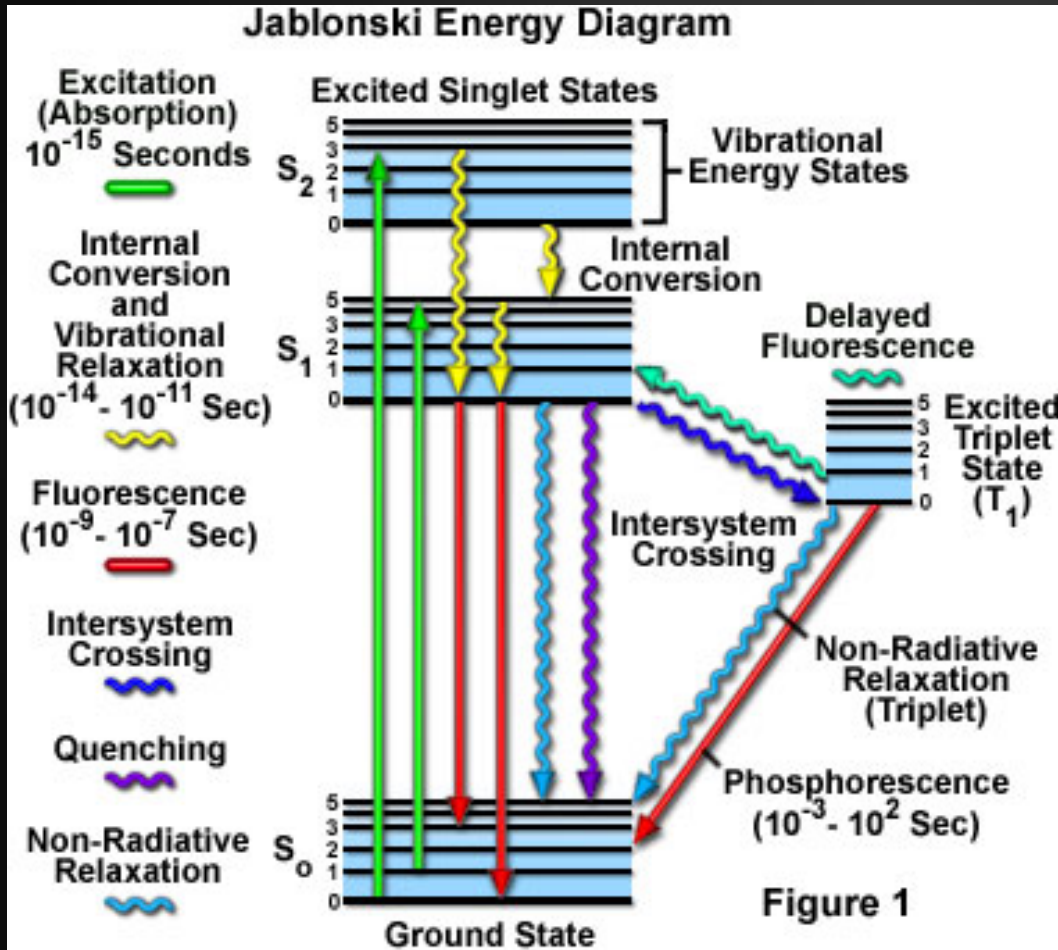
Unlike fluorescence, a phosphorescent material does not immediately emit light.

Absorbed radiation may be re-emitted for up to several hours after original excitation.

(wikipedia ;))



Fluorescence basics



<http://www.olympusmicro.com/primer/java/jablonski/jabintro/>

10^{-15} s : 1 femto sec

10^{-14} s : 10 fs

10^{-13} s : 100 fs

10^{-12} s : 1 pico sec

10^{-11} s : 10 ps

10^{-10} s : 100 ps

10^{-9} s : 1 nano sec

10^{-8} s : 10 ns

10^{-7} s : 100 ns

10^{-6} s : 1 micro sec

10^{-5} s : 10 μ s

10^{-4} s : 100 μ s

10^{-3} s : 1 milli sec

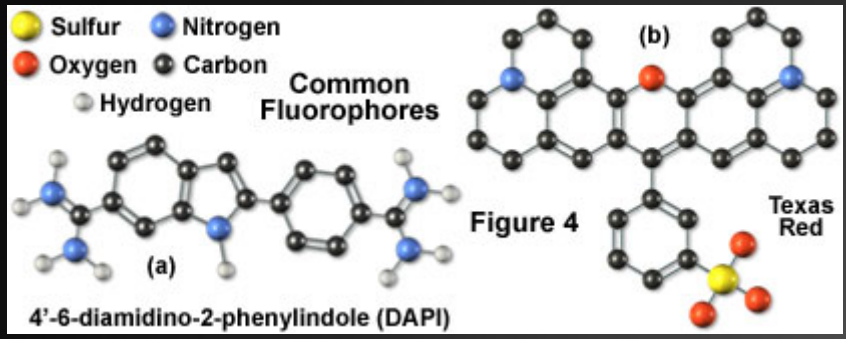
10^{-2} s : 10 ms

10^{-1} s : 100 ms

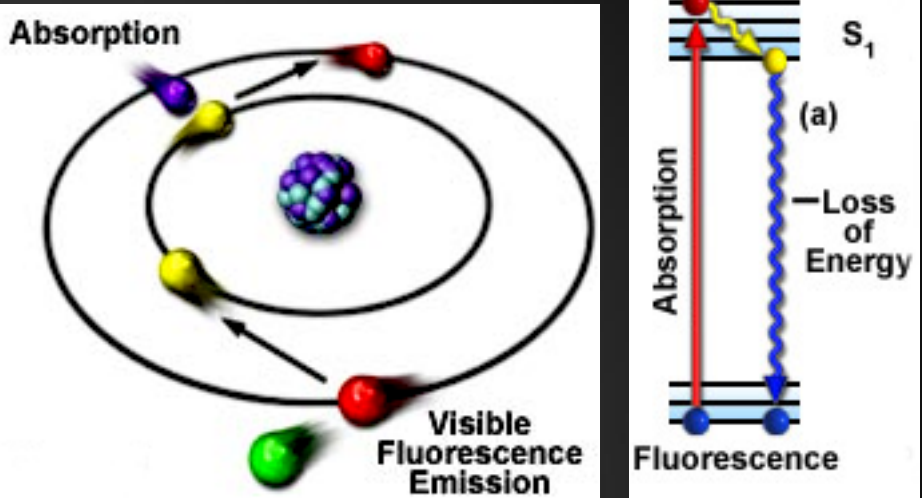
10^0 s : 1 second

Fluorescence basics

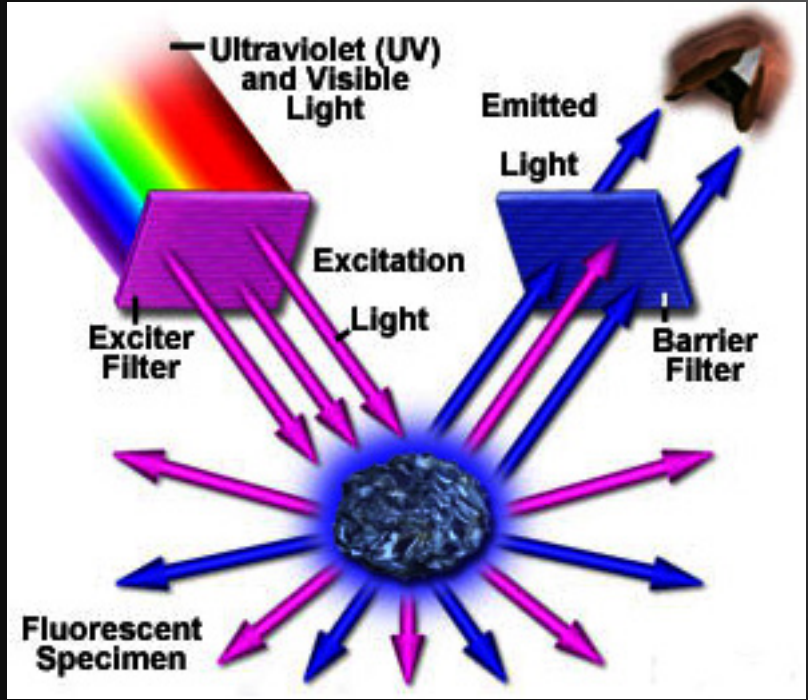
Examples of fluorescent probes



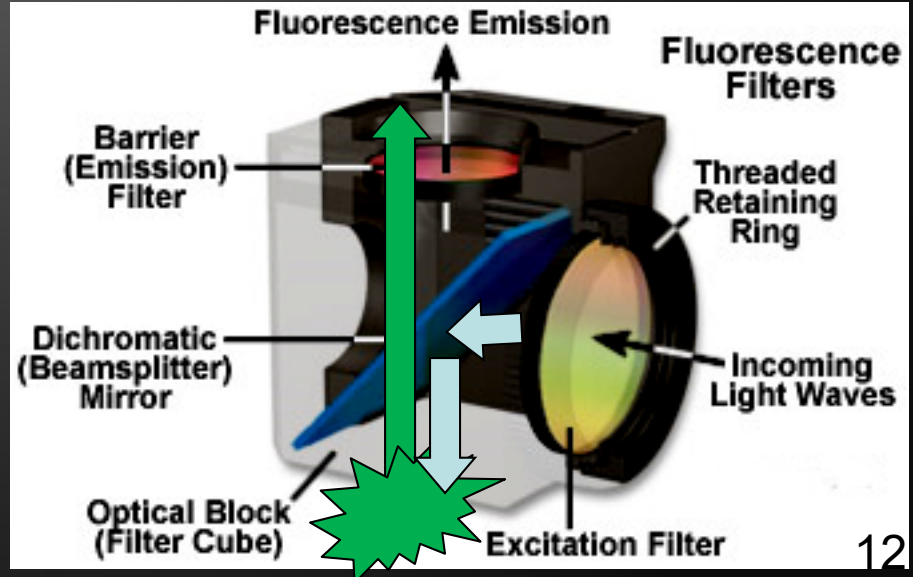
Principle of fluorescence



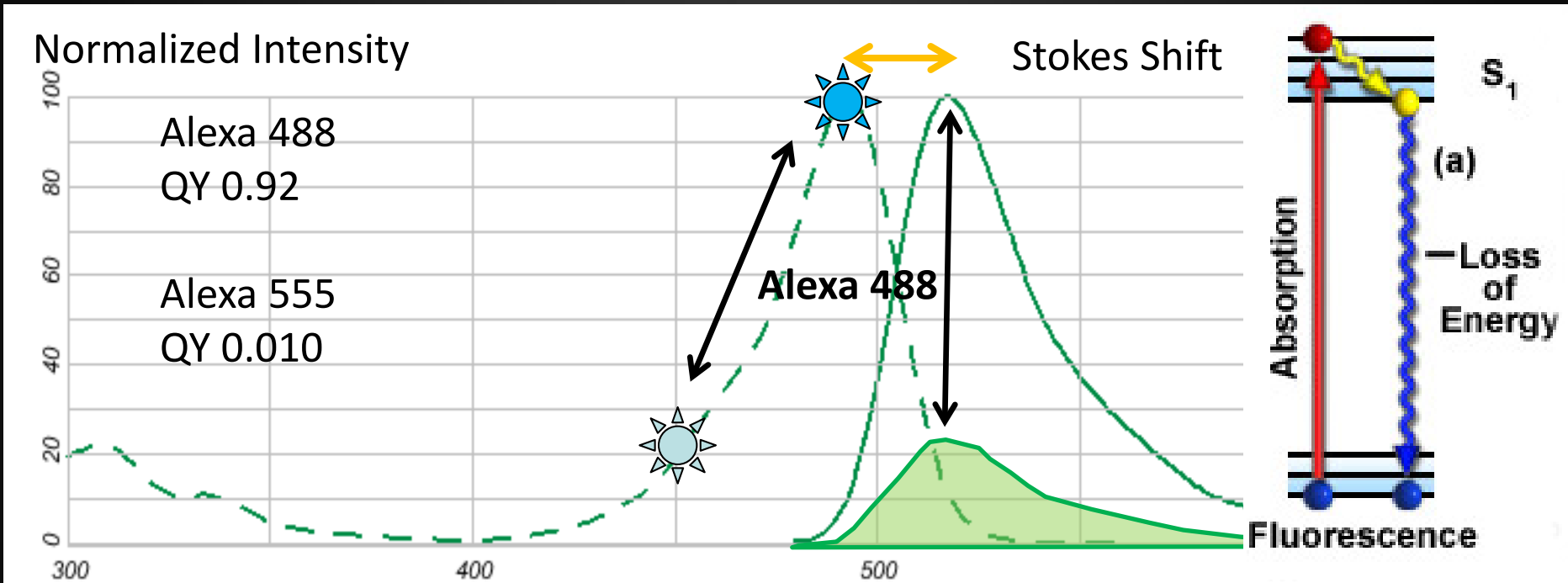
Principle of fluorescent microscope



Excitation-Emission filter cube



Fluorescence: The spectra



X axis: λ in nm

Y-axis: Intensity or probability of event that A) fluorophore absorbs the light for excitation (dashed line) and B) Fluorophore emits light (full line)

- Ex peak at 100% \rightarrow em peak at 100%, ex 20% \rightarrow em 20%, same range of emission
- Stokes shift: gap between ex-peak and em peak = (loss of energy, dissipation)
 \rightarrow important for separation of excitation and emission light in microscope etc

Other important features of fluorophores:

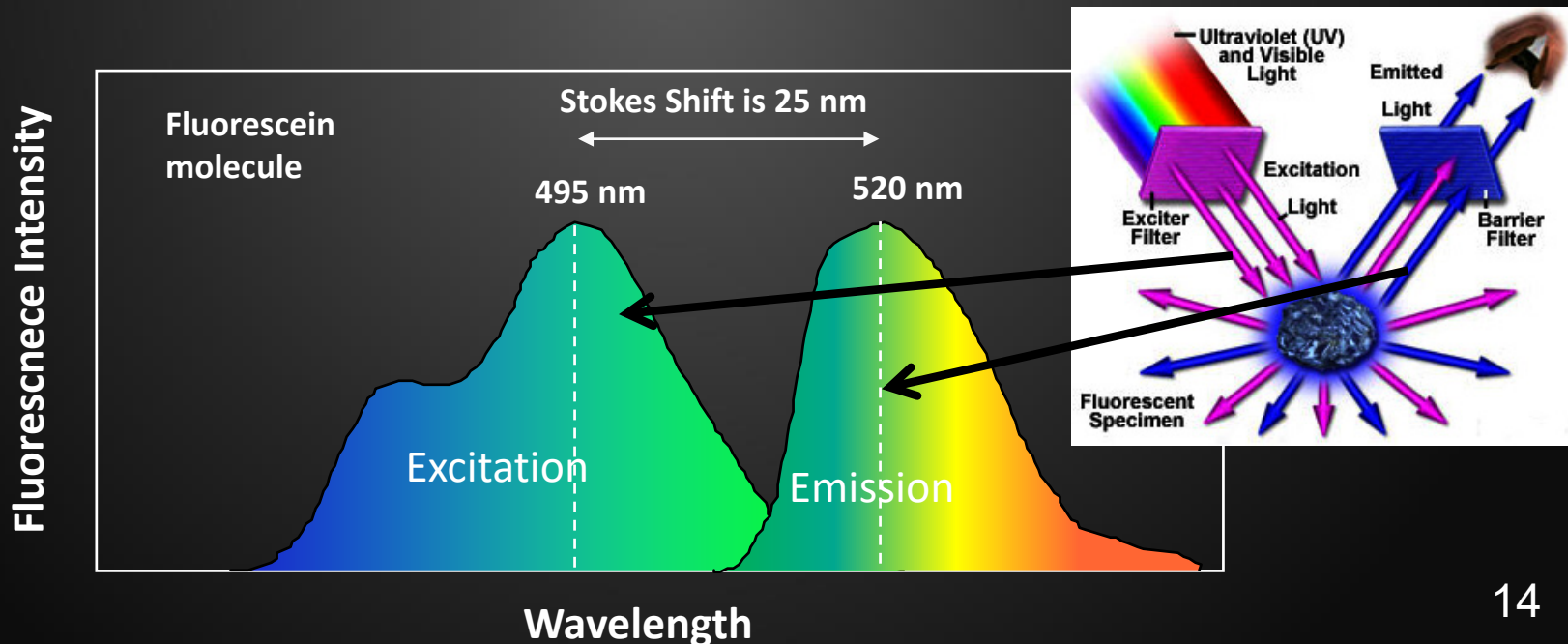
- Extinction coefficient: absorption efficiency of a photon at particular wavelength
- Quantum yield: proportion of photons emitted at λ_{em} to those absorbed at λ_{ex}

Stokes shift

Stokes shift is the energy difference between the lowest energy peak of absorbance and the highest energy of emission.

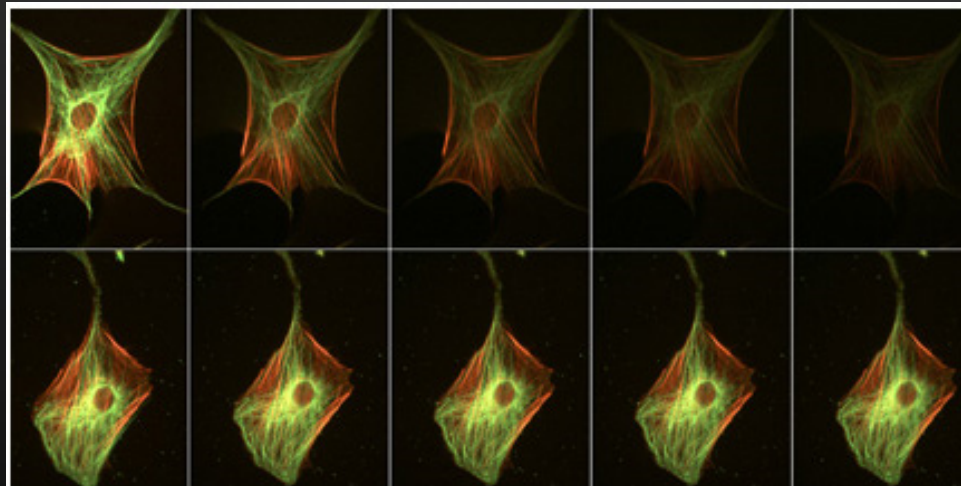
The Stokes shift is an extra for observing fluorescence; without it there would be (almost) no way to distinguish between excitation and emitted light

Probes with varying Stokes shifts are very useful for multicolor applications

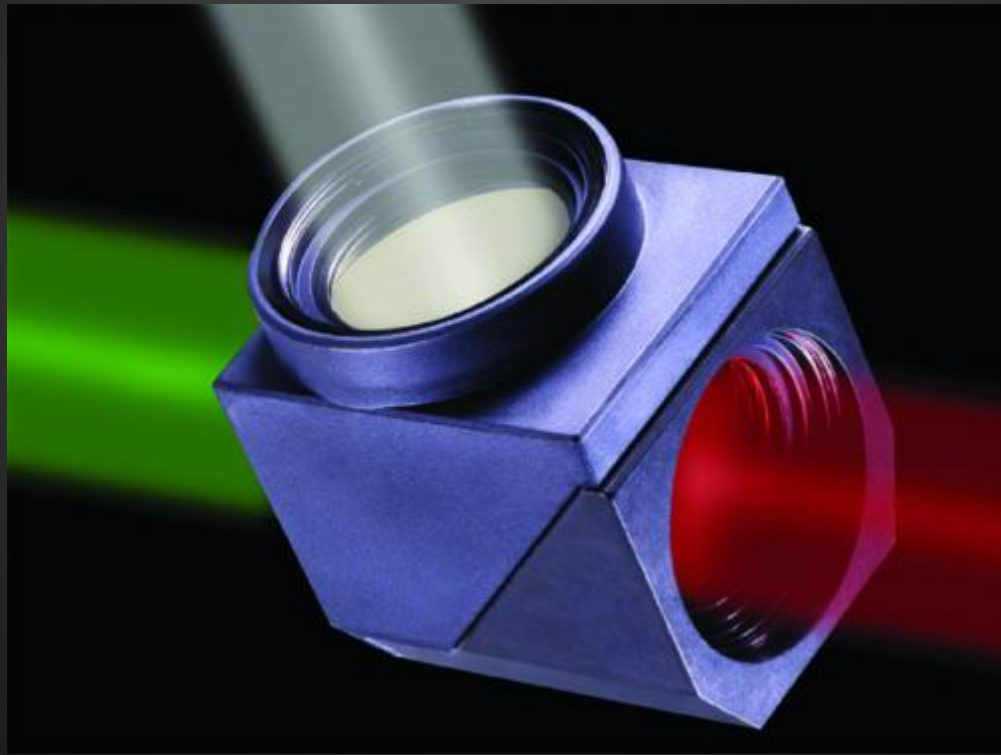


Photobleaching, quenching

- Defined as the irreversible destruction of an excited fluorophore
- Methods for countering photobleaching
 - Illuminate for shorter times
 - Use high magnification, high NA objective
 - Use wide emission filters – more signal to capture (may create problems with multiple probes)
 - Reduce excitation intensity
 - Use “antifade” reagents (not compatible with viable cells)



Filter and dichroic mirror



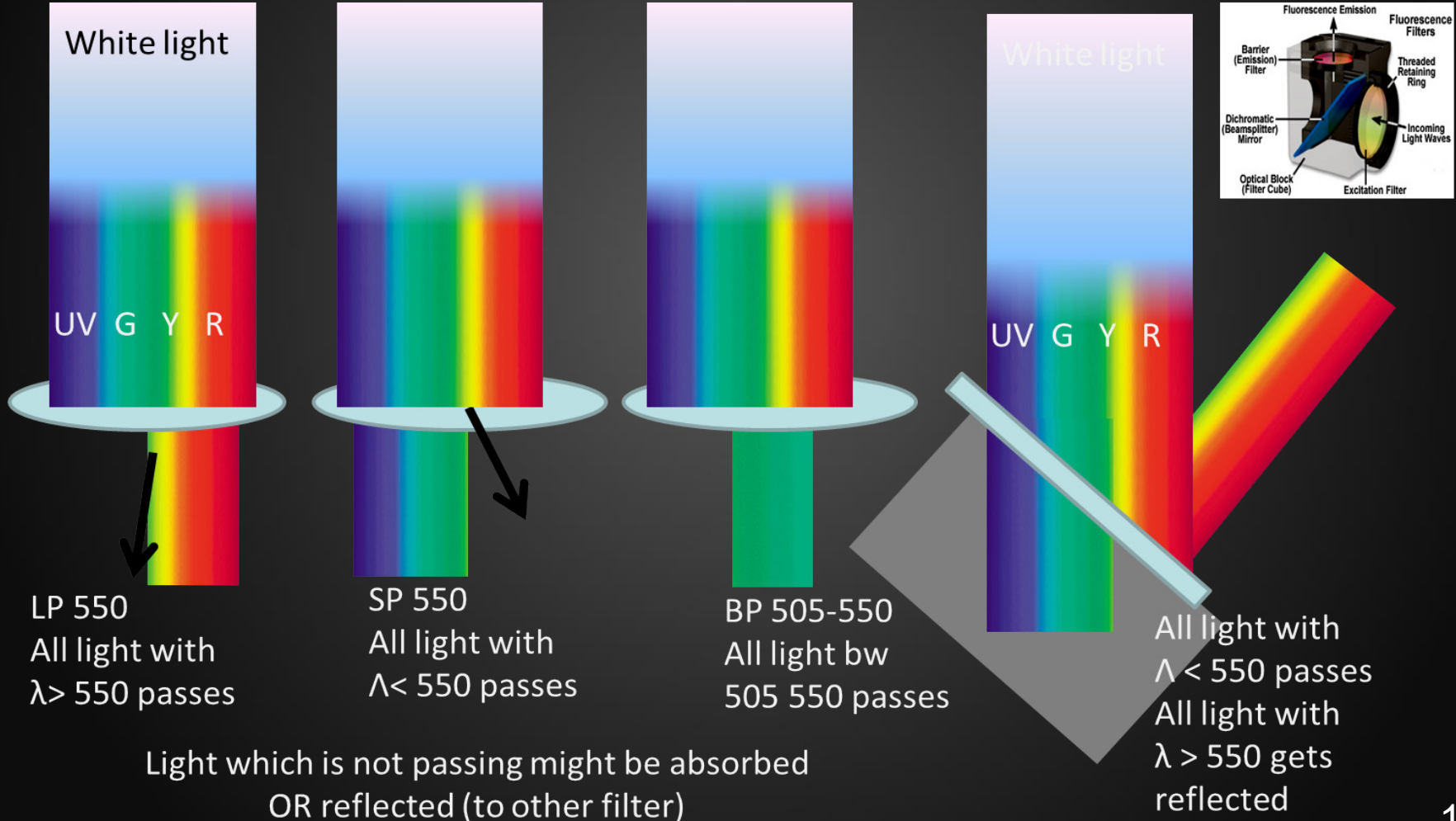
Filter and dichroic mirror

Longpass
LP filter

Shortpass
SP filter

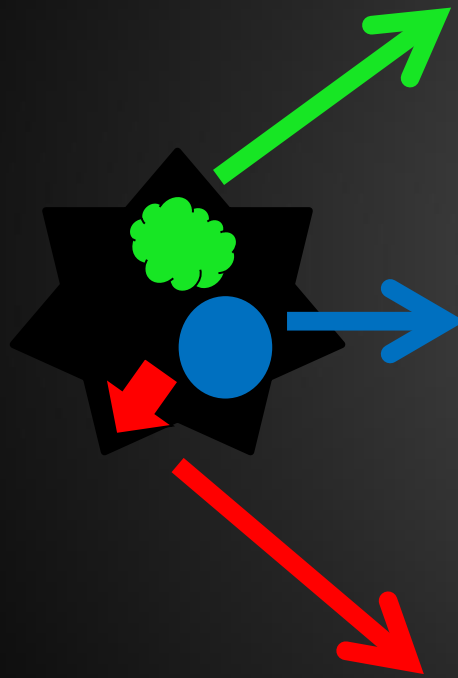
Bandpass
BP filter

Dichroic mirror, Beamsplitter
FT 550

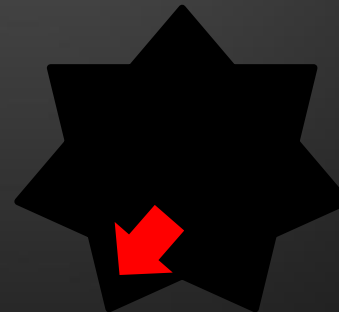
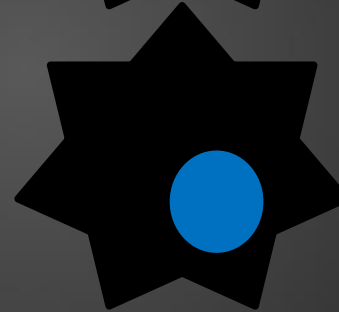
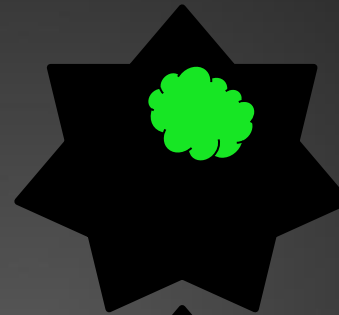


Dealing with fluorescence

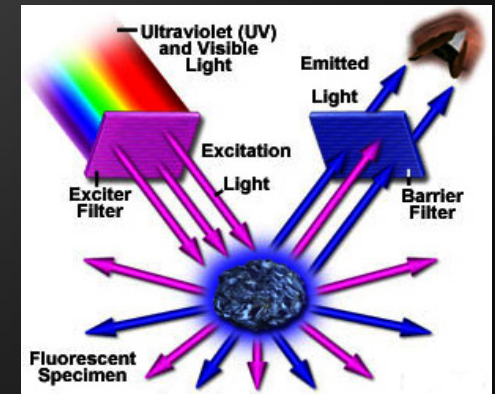
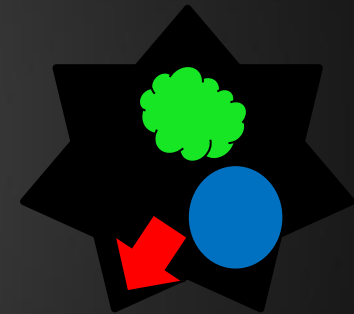
Cell sample



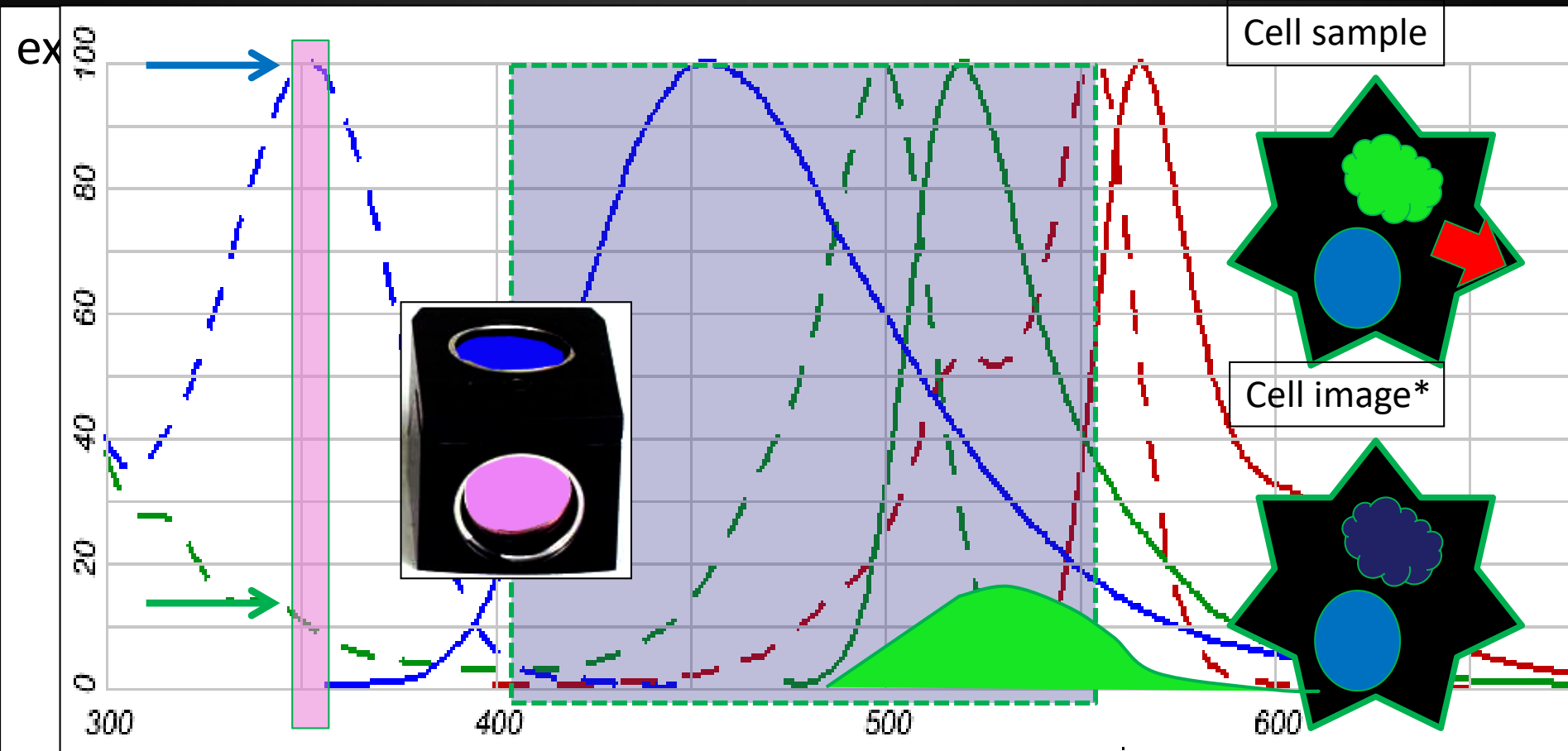
Cell images



merged RGB image

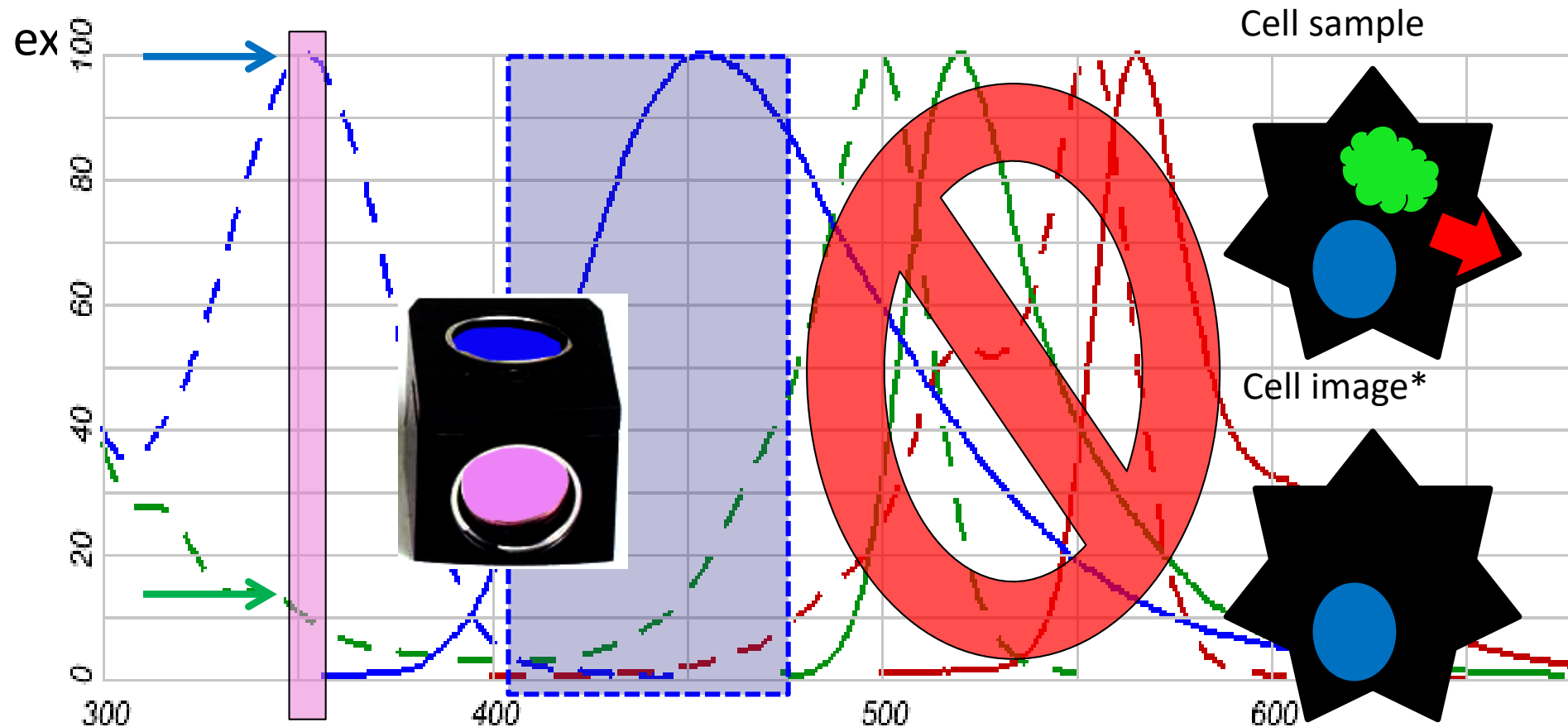


Dealing with fluorescence



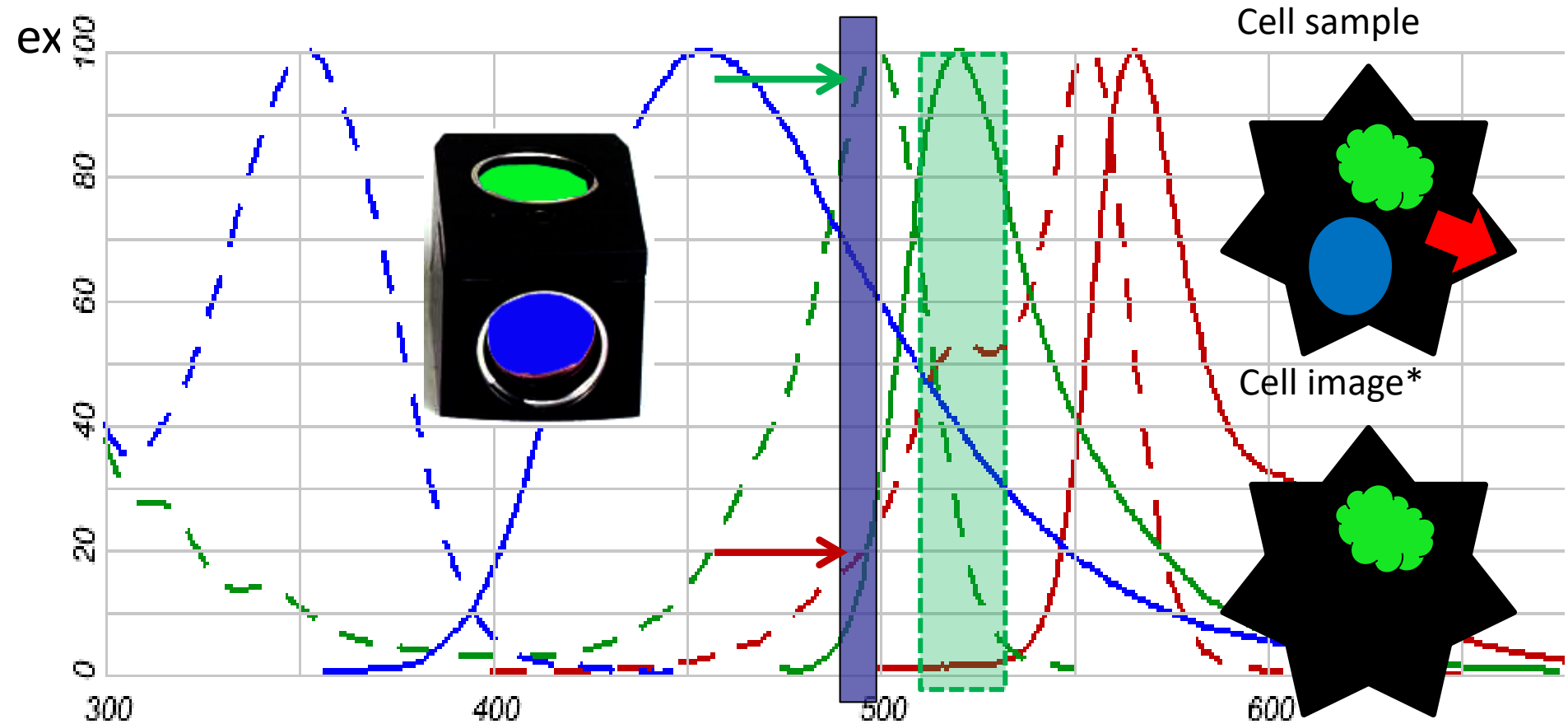
Excitation 350 nm excites Blue and Green, using BP filter 400-550 collects them both. *Remember: the camera is color blind. You decide with your choice of filter what it will see.

Dealing with fluorescence



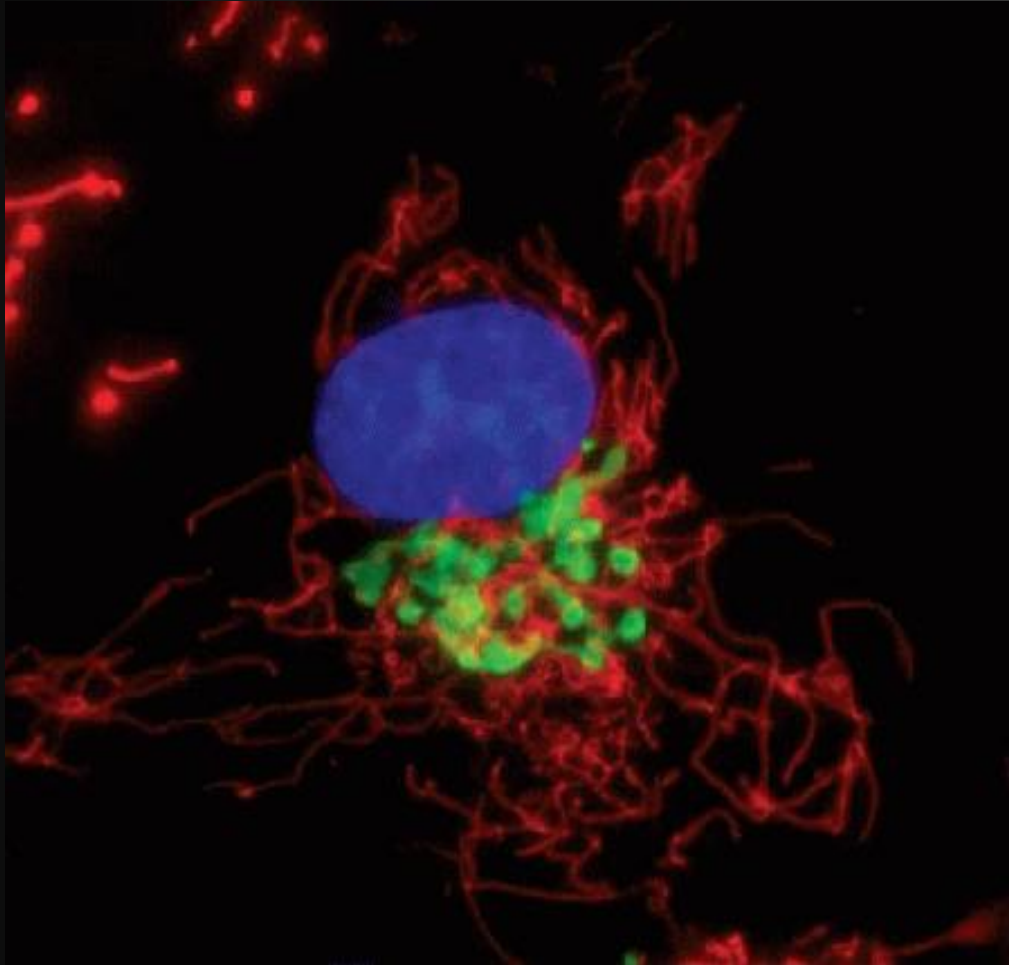
Excitation 350 nm excites Blue and Green, using BP filter 400-480 collects only the blue. *Remember: the camera is color blind. You decide with your choice of filter what it will see.

Dealing with fluorescence



Excitation 480 nm excites Green and Red, using BP filter 510-530 collects only the green. *Remember: the camera is color blind. You decide with your choice of filter what it will see.

Specific Organelle Probes



Gibco® human aortic smooth muscle cells (HASMC, Cat. No. C0075C) were transduced with CellLight™ Golgi-GFP , CellLight™ Mitochondria-RFP Hoechst 33342

Imaging was performed on live cells using a DeltaVision® Core microscope + standard DAPI/FITC/TRITC filter sets.

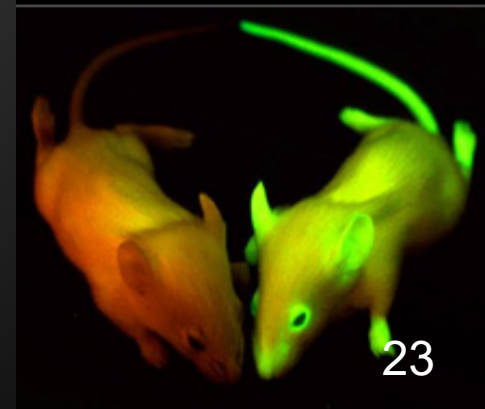
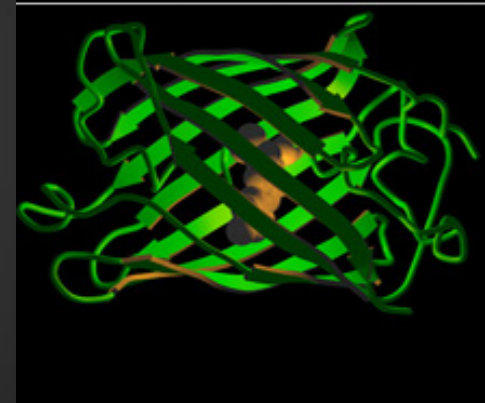
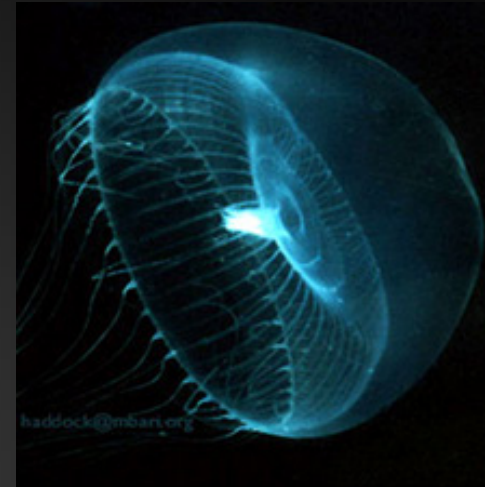
Fluorescent protein

- GFP - *Green Fluorescent Protein*
 - GFP is from the chemiluminescent jellyfish *Aequorea victoria*
 - excitation maxima at 395 and 470 nm (quantum efficiency is 0.8) Peak emission at 509 nm
 - contains a p-hydroxybenzylidene-imidazolone chromophore generated by oxidation of the Ser-Tyr-Gly at positions 65-67 of the primary sequence
 - Very stable
 - Major application is as a reporter gene for assay of promoter activity
 - requires no added substrates
 - Now in the enhanced form of eGFP , eYFP, eCFP

<http://gfp.conncoll.edu/>

<http://brainwindows.wordpress.com/category/gfp>

<http://www.biojobblog.net/2008/10/08/gfp-finally-gets-its-due/>



Important points

- Fluorescence is the primary information source for confocal microscopes and flow cytometry equipment
- Fluorescence emission is longer than the exciting wavelength
- Dye molecules must be close to, but below saturation levels for optimum emission
- Fluorescence probes must be appropriate for the excitation source and the sample of interest
- Correct optical filters must be used for multiple color fluorescence emission

THANKS FOR YOUR ATTENTION!

